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## Covalent bonds in superconducting rapid cooled alloys: (Bi<sub>1.7</sub>Pb<sub>0.3</sub>Sr<sub>2</sub>Ca<sub>n-1</sub>Cu<sub>n</sub>O<sub>2n+4+δ</sub>)<sub>2</sub>, n=1 to 9 detected by local atomic enhanced XRD

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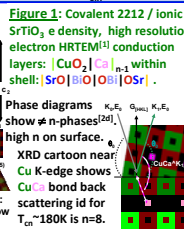
# Covalent bonds in superconducting rapid cooled alloys: $(\text{Bi}_{1-x}\text{Pb}_x\text{Sr}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+4})_2$ , $n=1$ to $9$ detected by local atomic enhanced XRD

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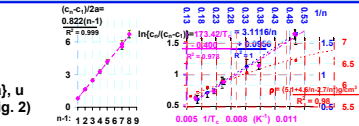


**Abstract:** High temperature layer superconducting cuprates (HTLSC) alloys:  $(\text{Bi}_{1-x}\text{Pb}_x\text{Sr}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+4})_2$ ,  $n=1-9$ , called  $[2; 2; n; 1; n]$  prepared in Uzbekistan from melts in concentrated sun flux, followed by rapid cooling, SFQA technology that preserves melt tiling by confinement of conduction layer:  $(\text{CuO}_2)_{1/2} \parallel (\text{CuO}_2)_{1/2} \parallel$  in the hard shell  $\parallel_1$  of the  $n=1$  phase,  $\parallel_1$   $[(\text{CuO}_2)_1 \text{Sr}(\text{BiO})_1 \text{Pb}_x \text{O}_2 \text{Sr}(\text{Pb}_x \text{O}_2 \text{Sr})_2 (\text{CuO}_2)_{1/2}]_n$ . Disproportion leads to different  $n$ -phase mixing, and periodic lattice distortions (PLD) through bond resonance covalent crystal axes:  $2l_{\text{Cu}} + 2r_{\text{O}} = 4.2\text{\AA} = 3.82\text{\AA} + d(\text{CuO}) + 2.32d(\text{O}) = 3.2\text{\AA} + c_{\text{Cu}} - c_{\text{O}}/2(n-1)$ . Individual phase local bonds **Cu-Ca**, **Cu-Sr**, **Cu-O-Cu** are ascertained by X-Ray diffraction near Cu K-edge followed by **Cu-Ca** bond scattering. The HTLSC density,  $\rho$  decrease as  $n$  increases obtains  $T_{c,n} \sim 180\text{K}$ , identified due to  $n=8$  phase on the surface of bulk  $n=4$ . Thus bond resonance leads to higher  $n$  phases on surface and predicts:  $T_{c,n} \rightarrow 298\text{K}$  as  $n \rightarrow 50$ , if the activity of skin depth is within  $c_{90}$  of a few  $\mu\text{m}$  even in mixed  $n$  phases.

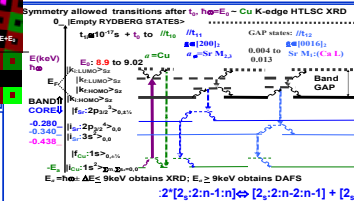
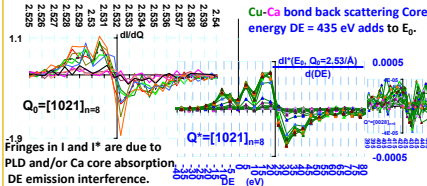
**Introduction:** Conductivity in HTLSC depends on  $\rho_{\text{electron}}$  grain boundaries, GB. Covalent vs ionic. 1. HRTM 2212: 1998 WY LIANG et al. 2. Non-bonding HTLSC Cu<sub>2</sub>O<sub>2</sub> HOMO:  $\rho_{\text{Cu}} > 10^{-10}$  show  $\neq n$ -phases<sup>[2d]</sup>, high  $n$  on surface. XRD cartoon near Cu K-edge shows  $\text{Cu}$  bond scattering for  $T_{c,n} \sim 180\text{K}$  is  $n=8$ .



**Figure 3:** Activity of free  $2e \leftrightarrow e_2^-$  in confined layers,  $\ln(V/V_{n=1}) = \ln(c_{\text{Cu}}/c_{\text{O}}) + 2e/n, 1/T_{c,n}$ .  $-\Delta G^*_{\text{Cu}}/k_B T = \ln(c_{\text{Cu}}/c_{\text{O}}) = 173.42/T_{c,n} - 0.4001$ .  $\{a, b, c_1, (c_{\text{Cu}} - c_{\text{O}})/2(n-1)\} = \{3.82\text{\AA}, a_{\text{Cu}}, 24.6\text{\AA}, 0.822\text{\AA}\}$ ,  $u = \text{PLD amplitude that gives rise to XRD structure (Fig. 2)}$



**RESULTS:** Figure 2:  $K_0$  incident X-rays at  $E_0 = 8940$  to  $9025\text{eV}$  scatter to  $K_n$  obtain  $1\text{V} = Q_0 = K_0 - K_n$  compare to enhanced atomic XRD,  $1^* \text{ vs } Q^* = Q_0(1 + \Delta E/E_0 \cos(\text{Cu} \parallel \text{Ca} \parallel K_1))$  when  $K_0$  X-rays, at constant  $\theta$ ,  $1^*$  vs  $Q^* = \text{aligned for } Q_0 \text{ excite Cu } 1s$ . Bond along Cu-Ca path adds **Ca** core  $\text{DE} = 435\text{eV}$  to  $E_0$ , obtains  $Q^*$  Bragg reflection id of surface phase.



**Figure 4:** Energy level diagram for enhanced XRD in  $Q_0, Q^*$ . Cu-M bonds are identified by M core  $\Delta E$  energies (fig. 4) to be Cu-Sr, Cu-Ca, Cu-O-Cu in mixtures of  $n$  phases. The mixing of different  $n$ -phases has been considered a detriment to prepare HTLSC. But the formation of lower density, higher  $n$ -phases on the surface, within the skin depth suggest superconducting transport. In SFQA melts, disproportion mixes  $n$  phases:  $2^*[2; 2; n-1; n] \leftrightarrow [2; 2; n-2; n-1] + [2; 2; n; n+1]$ ,  $d\rho/dT = \Delta\rho_H(T, V)$ , (1)

Pressure,  $P$  waves are controlled by the Clapeyron relation (1). In XRD near the Cu K-edge, states  $a$  are excited by  $N$  h<sub>0</sub> photons at  $K_0$  incidence and scattered to  $K_1$ ,  $Q_0 = K_1 E_0 - K_0 E_0$ . But Cu-M bonds contribute through states  $a$ , core  $\Delta E$ , to obtain  $Q^* = K_1 E_{0+\Delta E} - K_0 E_{0+\Delta E}$  when:

$$0 |a\rangle_{\text{Cu}} \langle \text{HOMO} | + N \text{ h} \leftrightarrow (N-1) \text{ h} + \langle a(-1)s \rangle_{\text{Cu}} \langle \text{HOMO} | \text{LUMO} | \quad (2)$$

is followed in  $\leq 10^{-17}$ s by depletion of  $M a_s$  core states / emission / absorption / backscattering from  $|Q_{\text{HOMO}}\rangle$ .  $|Q^*\rangle_{\text{LUMO}}\rangle$ ,  $\theta_0, \theta_s \in Q^*$ . The Cu-M bond contributions are identified by DE

$\{HKL\}_n$  assignments identify  $n=1$  to 9 possible phases due to the degeneracy of reflections in the alloys. XRD reflections reveal chemical activity by the scattering paths along Cu-M bonds, also involved in the growth free enthalpy,  $\Delta G^*$ , which in turn also determine the alloy transport and superconductivity properties.

**Discussion:** 1. Resonance is indicated by the crystal axes average of purely ionic and covalent bonds. 2. The CuO non bonding Cu<sub>2</sub>O<sub>2</sub> HOMO indicates  $\{10L\}$  GB formation in rapid cooled sun flux melts leads to  $n$ -mixing. GB formation in oxides induce rotations that are detected by local Cu atomic enhanced XRD of pairs of  $Q_0, Q^*$  from mixed  $n$  phases formed in melts, with strong local Cu-Ca, Cu-Sr, Cu-O-Cu bonds that are impossible to determine by XAS in mixed phases. 3. Interaction between the extended electronic states,  $|Q_0\rangle, |Q^*\rangle$ , decrease Gibbs free enthalpy of formation,  $\Delta G^*$  which is related to  $T_{c,n}$  (fig. 3) and may be due to tiling in melts exposed to sun radiation.

**Conclusion:** The SFQA alloys grown by novel fast cooling, preserves melt tiling, mix different  $n$  phases, and show GB formation typical of the parent SrTiO<sub>3</sub> promise useful applications with increased  $T_{c,n} \sim 150$  to  $180\text{K}$  in low density high  $n$ -phases formed on surface of alloys.

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