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**Fast melt cooled superconducting alloys:
(Bi_{1.97}Pb_{0.03}Sr₂Can-1CunO_{2n+4+δ})_{2 n/n' n<24} >intergrowth**

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Fast melt cooled superconducting cuprates: $(\text{Bi}_{1.97}\text{Pb}_{0.03}\text{Sr}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+4+b})_2 n/n' n \leq 24$ intergrowth

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Abstract: High temperature layer superconducting cuprate (HTLSC) alloys: $(\text{Bi}_{1.97}\text{Pb}_{0.03}\text{Sr}_2\text{Ca}_{n-1}\text{Cu}_n\text{O}_{2n+4+b})_2$ called $(2_s; 2:n-1:n)$ have been grown from n-oxide stoichiometric melts in concentrated sun flux, followed by rapid cooling SFQA technology that preserves the melt tilting after annealing at $845 \pm 5^\circ\text{C}^{**}$. Synchrotron XRD at the DOE SLAC-SSRL near the Cu K-edge has identified the mixing of $n \neq n'$ alloys as observed by many in thin films last century. An ideal D^{17}_{4h} Space Group structure obtains $\{a_n, b_n, c_n\} = \{3.815 \text{ \AA}, a_n + u_n, 2d_p(n+3) + u_n\}$ where d_p is a perovskite sandwich, $(\text{CuO}_2)_{1/2}|\text{Ca}|(\text{CuO}_2)_{1/2}$ thickness, and u_n are the amplitudes of periodic lattice distortions, PLD also observed last century*. Many electron interactions lead to covalent bonds indicated by $2(r_{\text{Cu}^{2+}} + r_{\text{O}^{2-}}) = 4.2\text{ \AA} > a = 3.82\text{ \AA} > d(\text{CuO}) + 2^{-3/2}d(\text{O}_3)$. Thus n-nano clusters grow n-1 perovskite sandwiches confined within a hard shell $(\text{CuO}_2)_{1/2}|\text{OSr}|_{\text{Bi}_{1.97}\text{Pb}_{0.03}}|\text{OBI}_{1.97}\text{Pb}_{0.03}}|\text{SrO}|(\text{O}_2\text{Cu})_{1/2}$. Disproportion reactions produce $n' = Nn + 3(N-1)n$ clusters that are supported by n cluster pillars (fig. 1). The increase in the transition temperature to the superconducting zone was determined by axial-torsional vibration measurements in transverse magnetic fields that obtain $T_c \rightarrow 190\text{K}$ as n increases in ppm emergent regions***. n, n' alloy mixtures identify Cu-Ca, Cu-Sr, Cu-O-Cu strong bond scattering. Example shows that enhancement at $Q_y = [11(n+3)]_{n=(n+1)2} = 2.53/\text{\AA}$ is also detected at $Q^* = [116]_{12}, [1029]_{12}$ near Cu K-edge due to Cu-M bond back scattering that identify $M = \text{Ca, Sr, O, Cu}$ (Table I, fig. 2).

Introduction: The quest is to ascertain how layer cuprates' T_c depends on chemical reactions in the melt subject to thermodynamics.

Our aim is to examine problems indicated by high resolution electron microscopy when the Gibbs free enthalpy of formation, ΔG_n^f governs the mixed n, n' layer growth, grain boundaries, GB (fig. 1) by using SFQA green technology methods to grow ceramics with low density, high T_c n'-n-alloys produced by chemical disproportions lead to pressure waves subject to the Clapeyron relation, e.g., $2'[2_s; 2:n-1:n] \leftrightarrow [2_s; 2:n-2:n-1] + [2_s; 2:n:n+1]$, where $dP/dT = \Delta H_f / (T \Delta V_f)$ produces GB, PLD, cages, stabilizing polarized states: $2e \leftrightarrow e_2^+$.

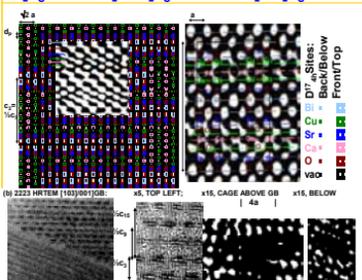


Figure 1: HRTEM^[1] has identified intergrowth, and GB cages where polarons, bi-polarons $2e \leftrightarrow e_2^+$ can catalyze transitions.

System	Center of G	h	k	l	h	k	l	h	k	l	h	k	l	h	k	l	h	k	l
BiO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
PbO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
SrO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
CaO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
CuO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
O	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00

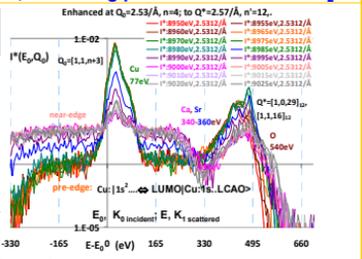


Figure 2: Enhanced XRD, shows that with K_0, E_0, Q_0 constant Cu 1s absorption, followed by scattering along Cu-M bond axes e.g., $\Delta E_{Cu} = 438\text{eV}$. ¹I recorded vs $E = E_0 + \Delta E$ obtains $Q^* = Q_0(1 + \Delta E/E_0) \cos(\text{CuM}^*K)$, $n, n' = 2$.

Table I n,n' coincidence enhanced XRD amplitudes used in analysis.

References: [1] W.Y. Liang, "Research Review, High Temperature Superconductivity" Cambridge University, (1998). [2] JG Chiginvadze et al., Phys. Lett. A, 300, 524 (2002); [3] DD Gulamova et al., Zh. Tekhn. Fiz. 79, 98 (2009) [4] JV Acrivos, Microchemical Journal 99, 239-245(2011)

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